Cornell Notes	Topic/Objective: Combined Gas Law		Name:				
			Class/Period: <b>Chemistry</b>				
AVID®  Decades of College Dreams			Date:				
Essential Question: What are the relationships between pressure, volume, and							
temperature of a gas sample?							
Questions:		Notes:					
How can volume,							
temperature,							
and pressure be							
related		Combined Gas Law Example Problem					
mathematically?		A gas at 100 kPa and 30.0° C fills a flexible container with an initial volume of					
		2.00 L. If the temperature is raised to 80.0° C and the pressure increased to 440					
		kPa, what is the new volume?					
Summary:							